



**Effects of climate change on sedimentary  
nitrogen cycling  
in the Bering Sea**

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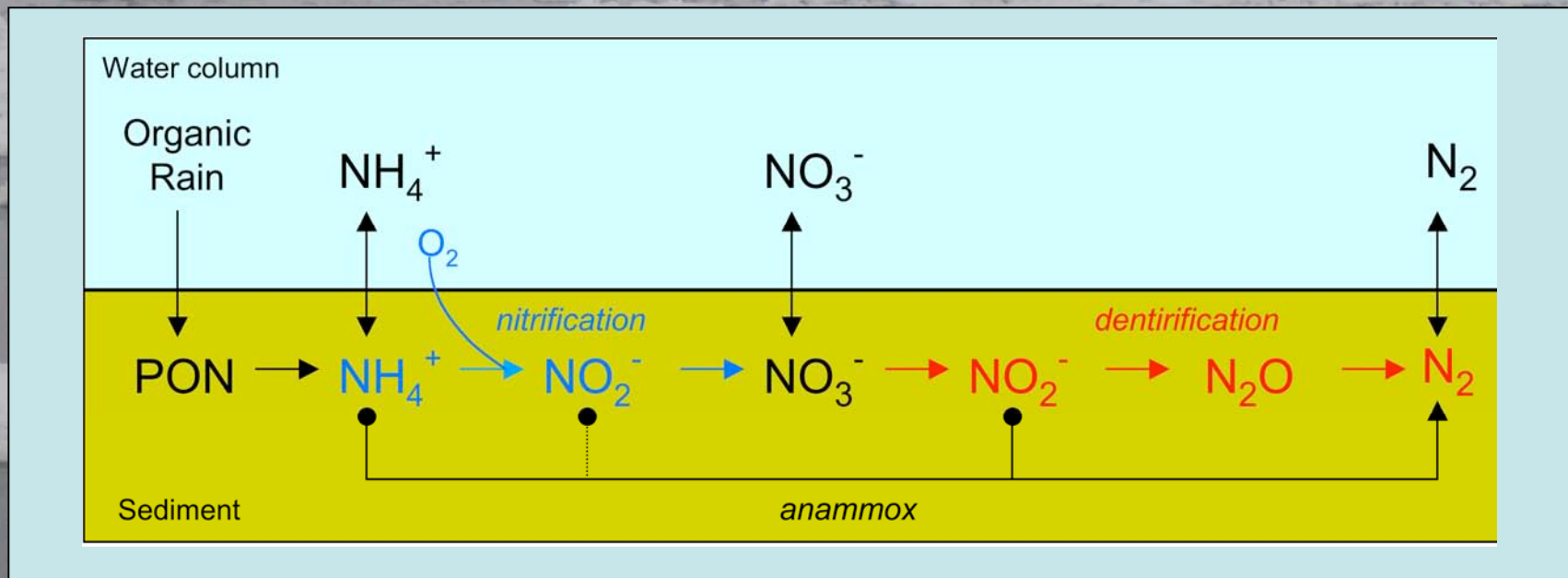
## Scientific Background

Rates of denitrification in shelf sediments of the southeastern Bering Sea in excess of  $1.1 \text{ mmol N m}^{-2} \text{ d}^{-1}$  are estimated to remove  $2.5 \times 10^{12} \text{ g}$  of reactive N per year.

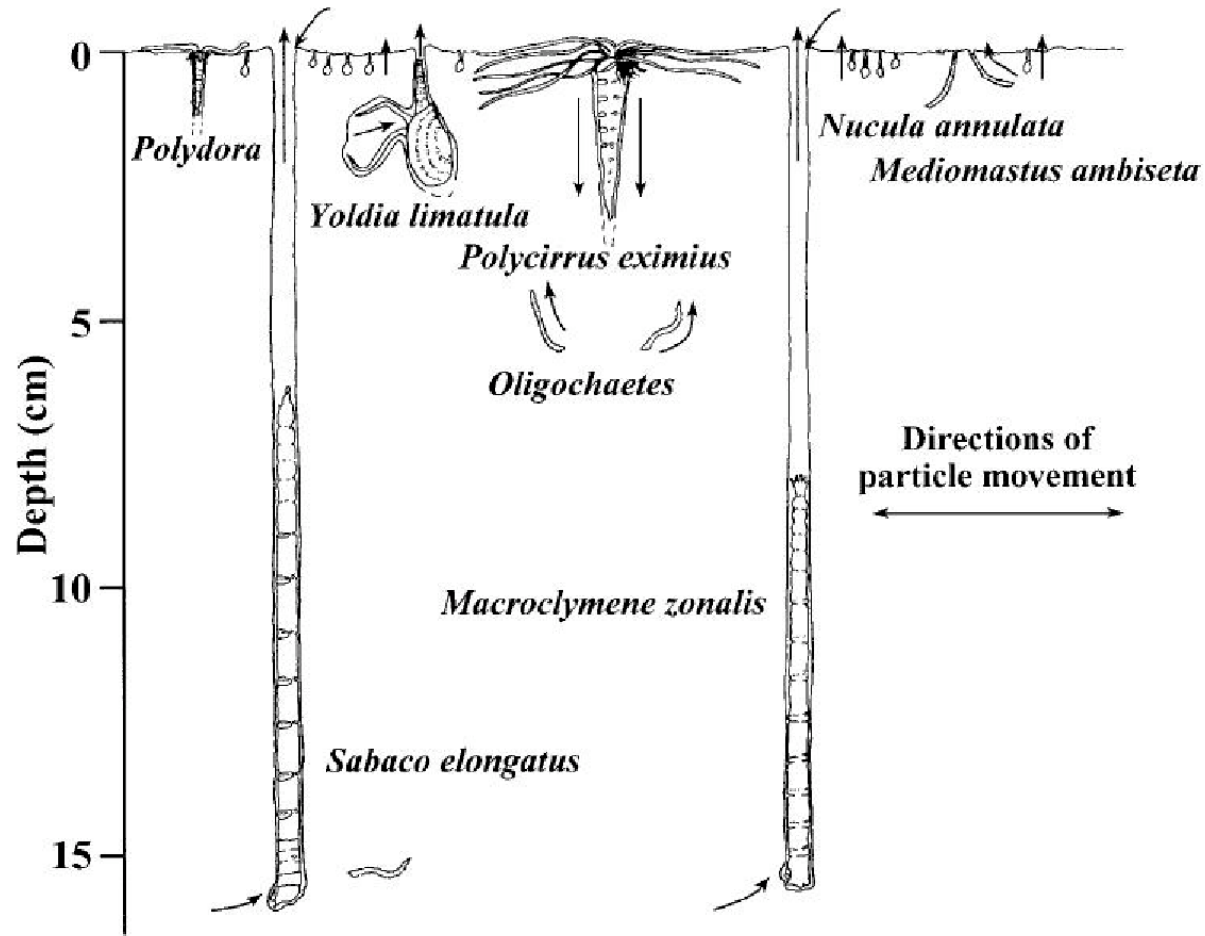
This amounts to about one third of the total nitrate supply to the Bering Shelf (Tomoyuki et al. 2004).

We will test the hypothesis that variation in the timing of the spring bloom will also change rates of denitrification in Bering Sea sediments, which will have substantial consequences for productivity in this region.

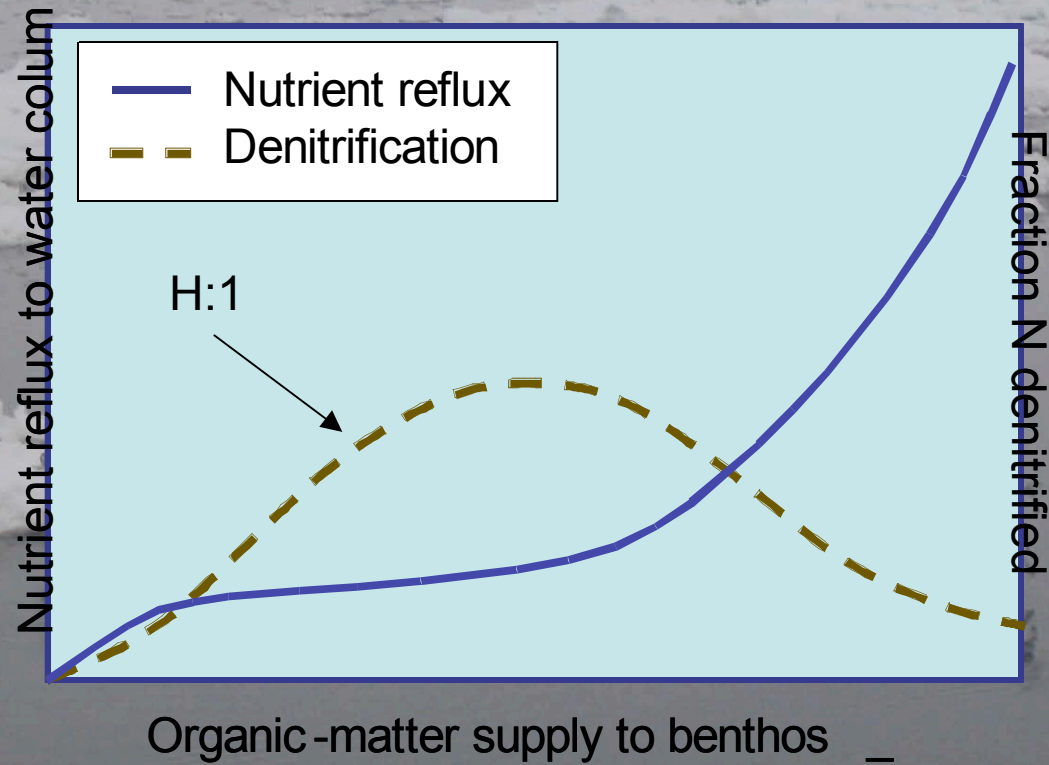
# Scientific Background – Sedimentary N-cycling



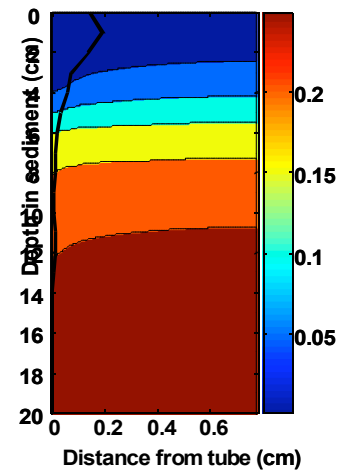
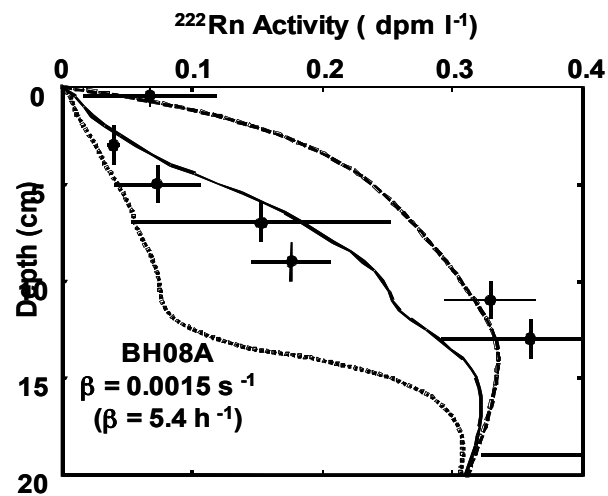
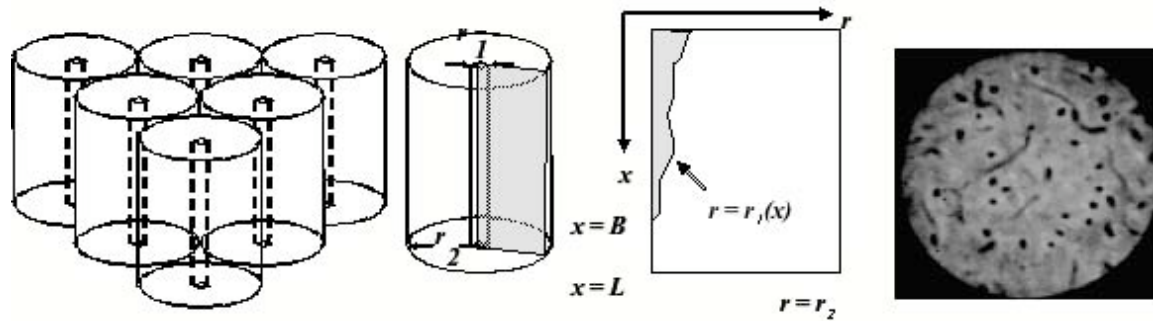
# Bioturbation complication:



H1: The fraction of fixed-N raining to sediments that is denitrified is a function of the rain rate and macrofaunal density



## H2: Macrofaunal irrigation is a function of the OC rain rate

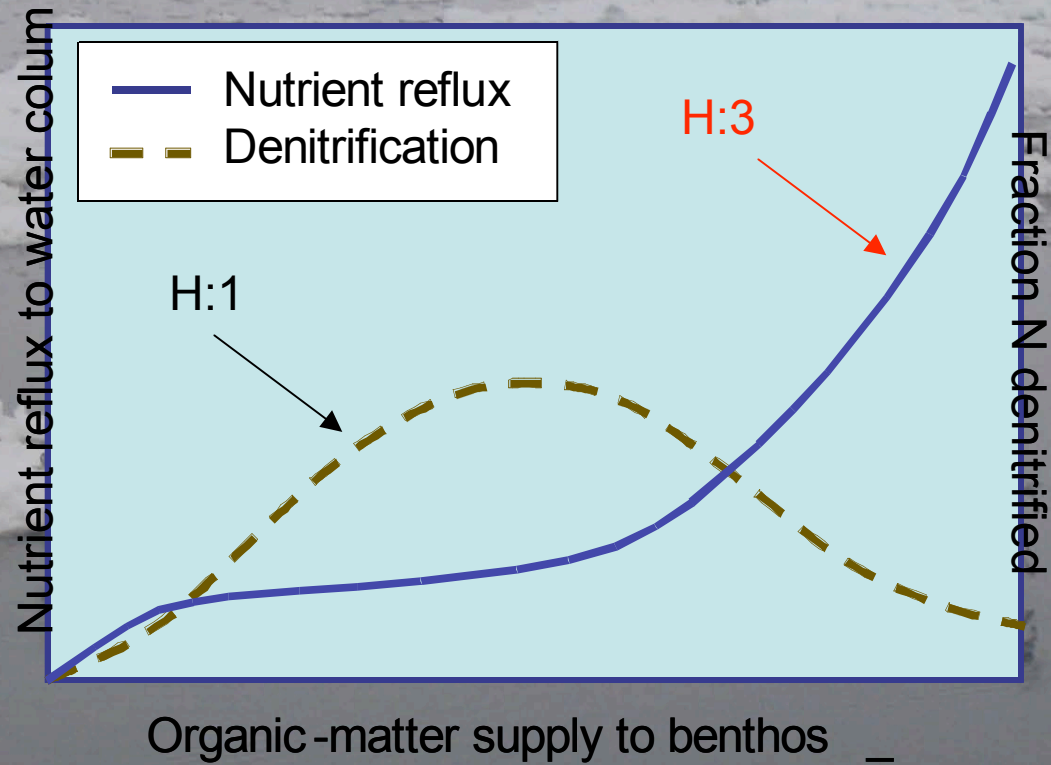


## Sediment Model:

$$\frac{\partial C_{x,r,t}}{\partial t} = \frac{\partial}{\partial x} \left( D'_x \frac{\partial C_{x,r,t}}{\partial x} \right) + \frac{1}{r} \frac{\partial}{\partial r} \left( r D'_x \frac{\partial C_{x,r,t}}{\partial r} \right) + \Sigma R_x$$

$$\frac{\partial C_{x,r,t}}{\partial t} = \frac{\partial}{\partial x} \left( D_x \frac{\partial C_{x,r,t}}{\partial x} \right) + \frac{1}{r} \frac{\partial}{\partial r} \left( r D_x \frac{\partial C_{x,r,t}}{\partial r} \right) + \Sigma R_x + \beta (C_0 - C_{x,r,t})$$

H3: Sediment reflux of fixed nitrogen ( $\text{NO}_3$  &  $\text{NH}_4$ ) to the overlying water will be a non-linear increasing function of organic supply to sediments



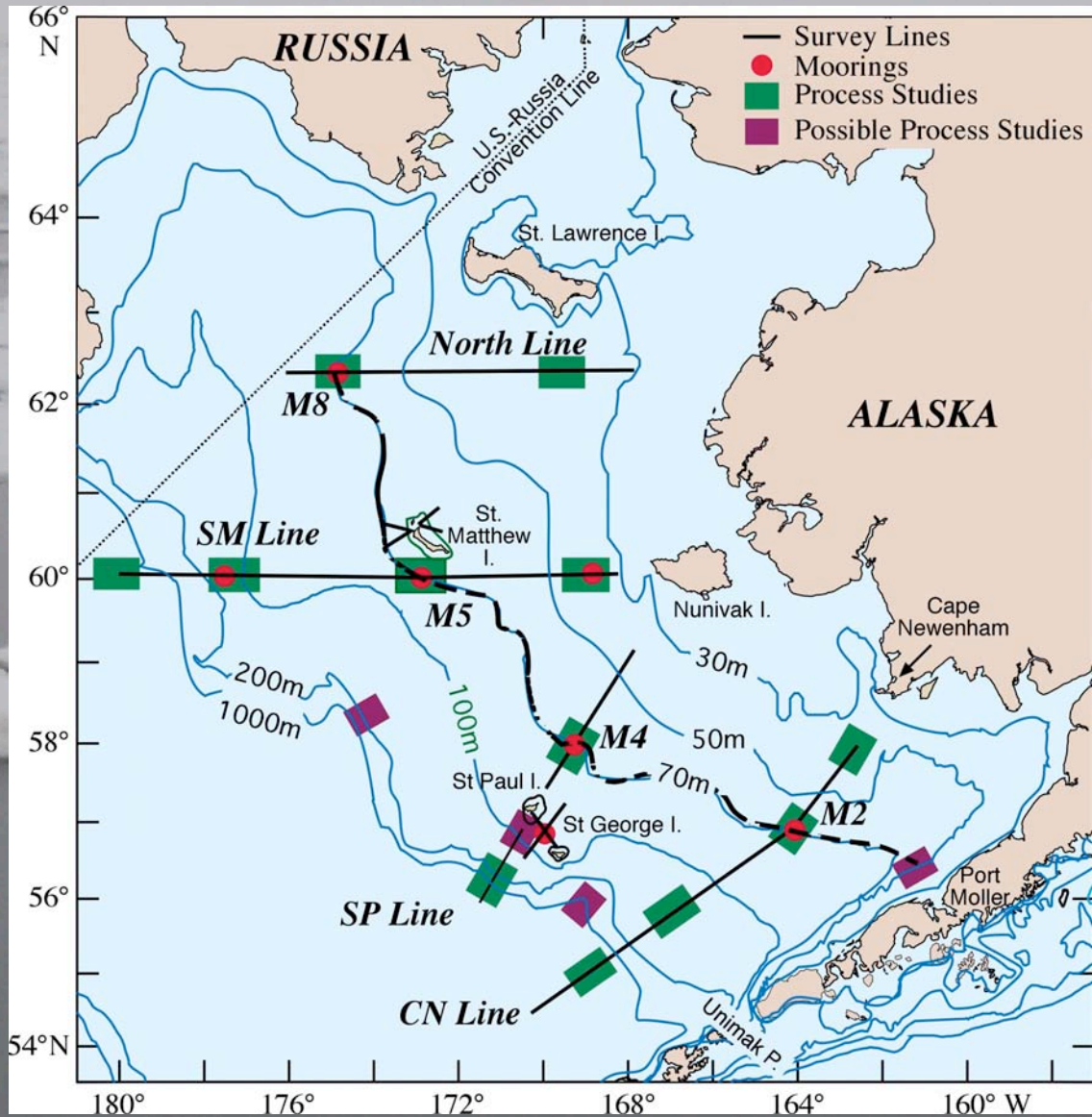




H4: Fraction of productivity hitting sea floor will change with timing of sea-ice melt

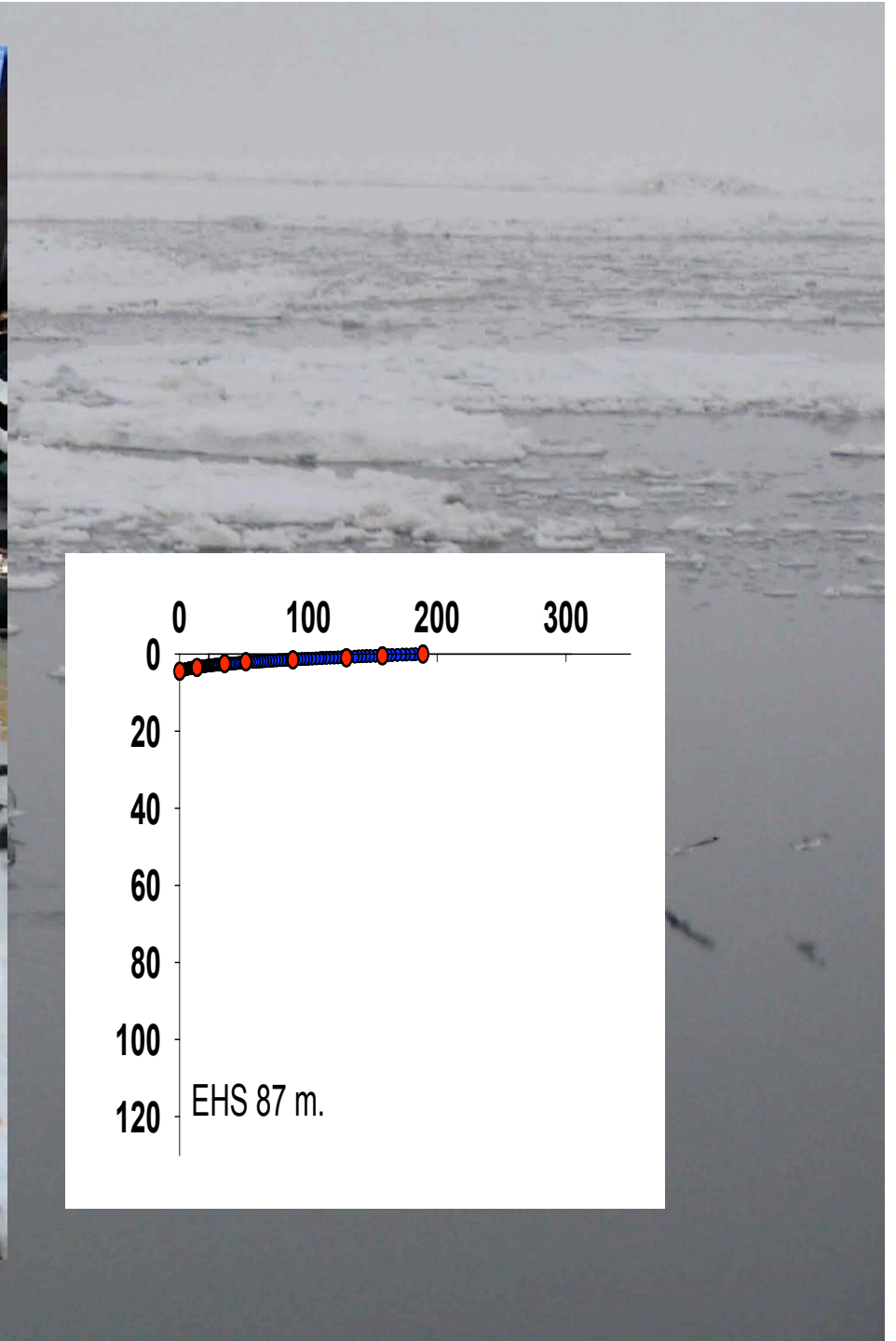
We will determine organic rain rate to bottom by measuring sediment carbon oxidation rate (via  $O_2$  flux, denitrification rate and sulfate reduction rate) and sedimentary OC burial ( $^{210}Pb$  sedimentation rate and OC content). Sedimentary carbon oxidation rate + burial = OC rain rate.

We will determine Productivity using the triple isotopes of oxygen method ( $^{18}O_2$ ,  $^{17}O_2$ ,  $^{16}O_2$ ) as well as gas exchange by  $^{222}Rn$  deficit.



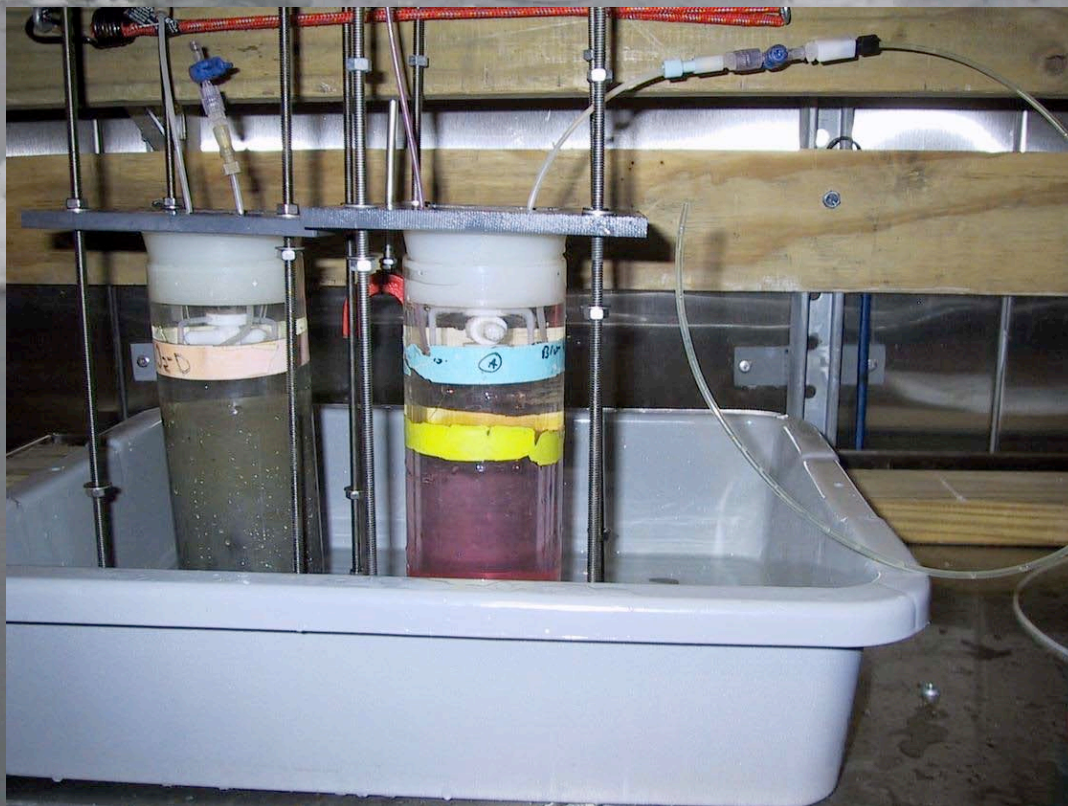
<u>Analysis</u>	<u>Measurement</u>	<u>Method</u>	<u>Instrumentation</u>	<u>Reference</u>
Flux cores (4 replicates)	O <sub>2</sub> /Ar, N <sub>2</sub> /Ar NH <sub>4</sub> NO <sub>3</sub> + NO <sub>2</sub>	MIMS Colorimetric Colorimetric	Quadrapole ALPKEM autoanalyzer ALPKEM autoanalyzer	Kana, 1998 Whitledge et al. 1981 Whitledge et al. 1981
Porewater profiles				
Squeezer	O <sub>2</sub> NO <sub>3</sub>	Amperametric NO <sub>3</sub> Colorimetric		Brandes and Devol, 1995 Brandes and Devol, 1995
Microelectrode	O <sub>2</sub> NO <sub>3</sub> + NO <sub>2</sub>	Amperametric Amperametric	UNISENSE electrode UNISENSE biosensor	Revsbech 1989
Sections (3 replicates)	O <sub>2</sub> NO <sub>3</sub> + NO <sub>2</sub> NH <sub>4</sub> pH HS	Amperametric Amperametric Colorimetric Electrode Ion-selective electrode	UNISENSE electrode UNISENSE biosensor UNISENSE biosensor ALPKEM autoanalyzer	Revsbech 1989 Whitledge et al. 1981
Anammox Sulfate Reduction	N-isotopes	incubation <sup>35</sup> SO <sub>4</sub> incubation	Finigan 253	Gilmour <i>et al.</i> , 1998 Kuypers et al., 2003 Fossing et al., 1989
Solid-phase				
Profiles (1 replicate)	<sup>210</sup> Pb <sup>222</sup> Rn	γ spectrometry α spectrometry Scintillation	Canberra GL 2820R Canberra 7401VR Applied techniques AC/DC-DRC-MK10-2	Shull 2001 Nittrouer et al. 1979 Mathieu et al. 1988
	TOC TN Porosity Grain size	CHN CHN Gravimetric Sieve, pipette	CE Elantech EA1112 CE Elantech EA1112	Hedges and Stern 1979
Burrow imaging	Animal burrows	CT-scanning	Picker 5000, St. Joseph's Hospital, Bellingham, WA	Shull and Yasuda 2001





## Sampling and Methods

- Whole core incubations (only in Spring 2004)
  - Incubated at in situ or near in situ temperatures
  - $N_2$  was measured in overlying water using membrane inlet mass spectrometry, MIMS



# Quadrupole mass spectrometer in *USCGC Healy* cold room

